STRUCTURAL STUDIES ON AVIDINORUBICIN, A NOVEL ANTHRACYCLINE WITH PLATELET AGGREGATION INHIBITORY ACTIVITY

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Avidinorubicin (MW 1,214, $C_{60}H_{86}N_4O_{22}$) was isolated from the cultured broth of strain NR0576 (which was identified as *Streptomyces avidinii* STAPLEY *et al.*) by butyl alcohol extraction, Sephadex LH-20 column chromatography and preparative HPLC. Avidinorubicin inhibited thrombin-induced platelet aggregation with an IC₅₀ being 7.9 μ M and was determined to be a novel anthracycline possessing two units of a new aminosugar, avidinosamine, in place of two decilonitrose groups in decilorubicin.

In the course of our screening for unique microbial products with pharmacological activity, avidinorubicin (1) that exhibited platelet aggregation inhibitory activity was isolated from the cultured broth of *Streptomyces avidinii*. Avidinorubicin was considered to belong to the nogalamycin¹⁾ group based on the presence of a unique anomeric proton in the ¹H NMR spectrum. After spectral analyses, avidinorubicin was determined to be a novel anthracycline containing two units of a new aminosugar, avidinosamine. This paper describes the taxonomy, fermentation, isolation, physico-chemical properties and structure elucidation of avidinorubicin.

Fermentation

The producing strain of avidinorubicin, NR0576, was isolated from a soil sample collected at Komono-cho, Mie Prefecture, Japan and identified as *S. avidinii* STAPLEY, MATA, MILLER, DEMNY and WOODRUFF²¹. A spore suspension of this strain (0.2 ml) was inoculated into a 500-ml baffled Erlenmeyer flask containing 100 ml of a medium consisting of glucose 2.0%, potato starch 2.0%, yeast extract 0.5%, toasted soya 2.0%, NaCl 0.25%, ZnSO₄ · 7H₂O 0.005%, CuSO₄ · 5H₂O 0.0005%, MnCl₂ · 4H₂O 0.0005%, CaCO₃ 0.32% and Nissan Disfoam 0.05%. The medium was adjusted to pH 7.0 before addition of CaCO₃. The inoculated flask was incubated on a rotary shaker at 27°C for 3 days at 190 rpm to make a seed culture. The resultant seed culture (12 ml) was transferred into a 3-liter baffled Erlenmeyer flask containing 600 ml of the same medium followed by incubation on a rotary shaker at 27°C for 3 days at 100 rpm. This seed culture (600 ml) was transferred into a 200-liter jar fermenter containing 120 liters of the same medium. The cultivation was carried out at 27°C for 6 days, agitated at 280 rpm and aerated at 150 liters per minute.

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Isolation Procedure of Avidinorubicin

The isolation procedure for avidinorubicin (1) is outlined in Fig. 1. The inhibitor in the broth filtrate (250 liters) was extracted with 162 liters of butyl alcohol at pH 8.5. The extract was concentrated to dryness *in vacuo* and suspended in 2.5 liters of H_2O . This solution was extracted with a mixture of chloroform and methanol (5:1, 2.5 liters × 3) and the organic layer was concentrated to dryness *in vacuo*. The residue (18.6 g) was chromatographed on a silica gel column (Wakogel C-200, 50 mm × 500 mm) and developed with a mixture of chloroform, methanol and aqueous ammonia (3:1:0.1) successively to yield a crude powder of 1 (554 mg).

Further purification was carried out by preparative HPLC with a Capcell pak C_{18} column (15 mm × 250 mm) developed with a mixture of methanol and 25 mM aqueous Na₂HPO₄ solution (3:1). The eluate containing 1 was concentrated to a small volume and extracted with butyl alcohol at pH 8.5. The organic layer was concentrated to dryness *in vacuo*. The residue was chromatographed on a column of Sephadex LH-20 with methanol to give a reddish powder of 1 (150 mg).

Structure of Avidinorubicin

The selected physico-chemical properties of 1 are summarized in Table 1, and its ¹H and UV spectra are shown in Figs. 2 and 3, respectively. The molecular formula of 1 was determined to be $C_{60}H_{86}N_4O_{22}$ based on the HRFAB mass (obsd 1,215.5911, +9.9 mmu for $C_{60}H_{87}N_4O_{22}$ (M+H)⁺) and NMR

Fig. 1. Isolation procedure of avidinorubicin.

Broth filtrate

adjusted to pH 8.5 extracted with BuOH

BuOH layer

concd under reduced pressure dissolved in H_2O extracted with $CHCl_3$ - MeOH (5:1)

Organic layer

concd under reduced pressure

SiO₂ column chromatography

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eluted with CHCl_3 - MeOH - aq NH_4OH (3:1:0.1)
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Preparative-HPLC
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Capcell pak C_{18} eluted with MeOH - aq Na₂HPO₄ (3:1) flow rate 8.0 ml/minute

Active fractions

concd under reduced pressure adjusted to pH 8.5 extracted with BuOH

BuOH layer

concd under reduced pressure dissolved in MeOH

Sephadex LH-20 column chromatography eluted with MeOH

Avidinorubicin

spectroscopic data. Its characteristic UV data suggested that 1 belonged to the anthracycline group, and the presence of a unique anomeric proton signal δ 6.50 strongly suggested that 1 was closely related to nogalamycin¹), decilorubicin³), arugomycin⁴) and viriplanin⁵). The latter three anthracyclines contain nitrosugars in the molecule as indicated by typical IR absorption at 1545 cm⁻¹ and 1350 cm⁻¹. The IR spectrum of 1 (Fig. 4) was almost identical with that of decilorubicin except

Table 1. Physico-chemical properties of avidinorubicin(1).

Appearence	Reddish powder
Molecular formula	C ₆₀ H ₈₆ N ₄ O ₂₂
MW	1,214
HRFAB-MS	Obsd 1,215.5911+9.9 mmu
	error for C ₆₀ H ₈₇ N ₄ O ₂₂
	$(M+H)^+$
UV λ_{max}^{MeOH} nm (ε)	237 (28,300), 255 (21,000),
inax ()	294 (6,200), 390 (3,000),
	498 (10,000)
$\lambda^{\text{MeOH + NaOH}}_{\text{max}}$ nm (ϵ)	238 (28,700), 254 (30,600),
max (7	294 (6,000), 366 (4,700),
	563 (11.000), 600 (10,500)
IR v_{max} (KBr) cm ⁻¹	3450, 1750, 1640, 1600, 1440,
max ()	1220, 1180, 1100~1000
MP	$206 \sim 209^{\circ}C$ (dec)
$[\alpha]_{D}^{24}$ (c 0.1, MeOH)	+ 523°



for the absence of the bands derived from nitro substituents. Consequently, we compared directly ¹H and ¹³C NMR spectra of 1 with those of decilorubicin. As shown in Table 2, ¹H and ¹³C chemical shifts of 1 are quite similar to those of decilorubicin except for the following protons and carbons adjacent to nitro group: I) C-3"-CH₃ (δ 2.00) and C-3"''-CH₃ (δ 1.92) proton signals of decilonitrose in decilorubicin are shifted upfield to δ 1.60 and 1.35, respectively, in 1; II) C-3" (δ 89.7)

Fig. 3. UV spectrum of avidinorubicin in MeOH.



and C-3^{'''} (δ 89.5) carbons attached to a nitro group are shifted to δ 52.2 and 52.2 in **1**. Taking into consideration of the chemical shifts of the 3^{''} and 3^{'''} positions and the molecular formula (C₆₀H₈₆N₄O₂₂ = decilorubicin - 2NO₂ + 2NH₂), the structure assignment of **1** is consistent with a compound with amino groups in place of the corresponding nitro groups in decilorubicin as shown in Fig. 5.

This assignment was confirmed by degradation experiments of 1 (Fig. 6). Hydrolysis of 1 (40% HCOOH, 80°C, 1 hour) gave 3 (MW 813, FAB mass 814 $(M+H)^+$, $[\alpha]_D^{24} + 412^\circ$ (c 0.1, MeOH), mp 138 ~ 142°C) and two sugar derivatives. The carbohydrate fragments were determined to be L-rhodosamine and 4-succinyl-L-diginose based on their NMR and FAB mass spectral data, and $[\alpha]_D$ values^{6,7)}.

The MW of 3 (HRFAB-MS, obsd 814.3760, -0.3 mmu for $C_{41}H_{56}N_3O_{14}$ (M+H)⁺), could easily be explained by cleavage of the rhodosamine and 4-succinyldiginose moieties from 1. Upon comparison of the ¹³C NMR spectrum of 1 with that of 3, glycosidation shifts were observed at C-7 (δ 69.1 for 1 vs. 62.6 for 3) and C-4''' (δ 85.4 for 1 vs. 77.5 for 3) in 1. These results indicated that the sugar units were removed from C-7 and C-4''' in 1 during hydrolysis.

Hydrogenolysis of 1 over Pd - BaSO₄ gave 2 (MW, 1,041) and rhodosamine. This result indicated that

	A	vidinorubicin		3	D	ecilorubicin
1	157.5		157.4		154.6	
2	139.7		139.1		ND	
3	131.2	7.75 (d, $J = 8.3$) ^a	131.0	7.75 (d, $J = 8.3$)	131.1	7.90 (d, $J=8$)
4	119.2	8.03 (d, J=8.3)	118.8	8.02 (d, J=8.3)	118.9	8.05 (d, $J=8$)
4a	134.8		134.6		ND	
5	181.8		186.7		ND	
5a	111.9		111.7		111.8	
6	154.5		154.5		157.4	
6a	138.0		ND		ND	
7	69.1	5.55 (br s)	62.6	5.50 (br s)	68.8	5.57 (br s)
8	41.2	2.65 (d, $J = 13$), 2.25°	41.5	2.52 (d, $J=13$), 2.01 (dd, $J=13$, 3)	41.7	$2.22^{\circ}, 2.65 \text{ (d, } J=13\text{)}$
9	67.4		68.2		67.5	
10	38.6	3.45 (d, $J = 18$), 2.85 ^b	38.7	3.56 (d, $J = 17$), 2.82 (d, $J = 17$)	38.5	2.88 (d, $J=18$), 3.47 (d, $J=18$)
10a	120.1		120.0		120.4	
11	157.0		156.5		156.2	
11a	111.2		111.2		111.4	
12	186.7		186.7		ND	
12a	137.2		ND		ND	
13	32.0	1.45 (s)	30.4	1.54 (s)	32.0	1.55 (s)
1'	99.0	6.50 (d, $J=3$)	98.8	6.43 (d, $J=3$)	98.8	6.46 (d, $J=3$)
2'	68.1	4.88 (dd, J=11, 3)	67.9	$4.84 (\mathrm{dd}, J = 11, 3)$	68.0	4.85 (dd, J=11, 3)
3'	61.6	3.34 ^b	61.4	3.29 (dd, J = 11, 1.5)	61.6	3.30 (dd, J=11, 2)
4′	80.7	4.70 (d, $J = 1.5$)	80.9	4.68 (d, $J = 1.5$)	81.3	4.70 (br d, $J=2$)
5'	78.0		78.1		77.8	
6′	23.2	1.88 (s)	23.2	1.87 (s)	21.5	1.78 (s)
$3' - N(CH_3)_2$	40.3	2.84 (s)	40.4	2.80 (s)	40.5	2.80 (s)
1"	100.9	5.58 (d, $J = 9.5$)	100.7	5.68 (br d, $J=9$)	100.2	5.90 (d, J=9)
2"	45.9	1.78 (dd, J = 14, 9.5),	45.4	2.27 (d, $J = 13$),	42.0	2.18 (dd, $J = 14, 9$),
2"	60.0	2.12		1.81 (dd, $J = 13, 9$)	00.7	2.9/(d, J = 14)
3" 4"	52.2	2 20 (1 7 10)	52.9	2.22 (1.7.10)	89.7	
4" 5"	87.9	3.20 (d, J = 10)	85.2	3.38 (d, J = 10)	82.3	3.78 (d, J = 10)
5	/0.8	4.20 (aq, J = 10, 6.5)	/0.0	4.30 (aq, J = 10, 6.5)	/0.2	4.21 (dq, J = 10, 6)
0 2″ CH	18.9	1.37 (d, J = 0.3)	18.7	1.43 (0, J=0.3)	18.4	1.33 (d, J=6)
3 -CH ₃	100.0	1.00(8) 5.27 (d. 1-0.5)	30.9	1.30(8) 5.35(4, 1, 0)	24.8	2.00 (S) 5.55 (A I , 0)
1	100.0	5.27 (0, J = 9.5) 1.65 (dd $I = 14, 0.5$)	99.8	3.35(0, J=9)	99.8 42.0	$5.55(0, J \approx 9)$
2	43.9	1.05 (dd, $J = 14, 9.5$), 1.95 (d, $J = 14$)	44.8	2.13 (d, J = 13), 1.71 (dd, $J = 13, 9$)	42.0	1.88 (dd, $J = 14, 9$), 2.56 (d, $J = 14$)
3‴	52.2		52.5	н. Н	89.5	
4‴	85.4	3.28 (d, J=10)	77.5	3.32 (d, J = 10)	83.8	3.66 (d, J=10)
5‴	70.8	4.13 (dq, $J = 10, 6.5$)	71.3	4.24 (dq, J = 10, 6.5)	70.2	4.18 (dq, J=10, 6)
6′′′	19.0	1.40 (d, $J = 6.5$)	19.1	1.58 (d, $J = 6.5$)	18.5	1.35 (d, $J=6$)
3 ^{′′′′} -CH ₃	32.0	1.35 (s)	31.5	1.40 (s)	24.8	1.92 (s)
1""	101.9	5.30 (br s)			102.1	5.35 (br s)
2""	29.2	ND			30.0	ND
3	73.4	ND			73.3	3.70 (ddd, J = 11.5, 6, 2.5)
4''''	71.8	5.77 (br s)			71.7	5.85 (br s)
5	66.6	4.28 (q, J=6.5)			66.8	4.53 (q, J=6)
ь р	F7.0	1.33 (d, J=6.5)			17.0	1.5/(d, J=6)
3 -OCH ₃	56.0	3.38 (S)			55.9	3.23 (s)
SucCO	173.3	2.004			173.4	ND
Suc CH_2	30.0	3.00°			30.0	
SucCH ₂ SucCOOH	30.0 175.4	5.00*			30.0 172.7	ND

Table 2. NMR chemical shifts of avidinorubicin, **3** and decilorubicin in pyridine- d_5 (δ ppm).

Table 2. (Continued)								
Avidinorubicin			3	Decilorubicin				
1'''''	100.9	5.27 (br s)		101.6	5.23 (br d, $J=3$)			
2'''''	32.0	1.30 ^b		30.0	2.20 ^ь			
3'''''	67.0	2.50 (m)		61.0	2.68 (m)			
4''''	67.4	3.85 (br s)		67.1	3.90 (br s)			
5'''''	68.1	4.48 (q, J=6.5)		68.0	4.46 (q, J=6)			
6'''''	17.8	1.60 (d, J=6.5)		17.6	1.59 (d, J=6)			
3"""-N(CH ₃) ₂	42.2	2.15 (s)		42.0	2.15 (s)			

Table 2. (Continued)

^a Multiplicity and coupling constant (Hz) are in parentheses.

^b J value could not be determined.

ND: Not determined.



Fig. 4. IR (KBr) spectrum of avidinorubicin.

Fig. 5. Structure of avidinorubicin (1).





Fig. 6. Degradation products of avidinorubicin.

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Decilorone (4)

rhodosamine was attached to C-7, and consequently 4-succinyldiginose was bound at C-4". The anomeric protons of the rhodosamine and diginose moieties in 1 (1""-H and 1""-H, respectively), observed at δ 5.27 and 5.30 as broad singlets (Table 2), indicated that these glycosidic bonds had α -configurations.

Methanolysis of 3 (5% HCl in methanol, 110°C, 18 hours) yielded a methyl glycoside of a novel aminosugar 5, (methyl α -L-avidinosaminide). 5 was determined to be methyl 3-amino-2,3,6-trideoxy-3-*C*-methyl- α -L-*ribo*-hexopyranoside based on its FAB mass ((M + H)⁺ 176) and ¹H and ¹³C NMR spectroscopic studies (Table 3). The equatorial orientation of the methyl group at C-3 was elucidated by analyzing ¹³C NMR chemical shifts and measuring NOEs. In most cases, the ¹³C NMR chemical shifts of 3-axial methyl

carbons of 3-C-methyl sugars are observed at higher field than those of 3-equatorial methyl carbons (Table 4)8). The ¹³C NMR chemical shift of the methyl group at C-3 in 5 is similar to those of equatorial methyls of other 3-C-methyl sugars (Table 4) and those of decilonitroses in decilorubicin (Table 2). In addition, NOEs were observed between 3"-CH₃ and 4"-H, and 3"'-CH₃ and 4"'-H in NOESY experiments of 3. Therefore, the 3-methyl substituent in 5 was determined to occupy an equatorial position. The absolute configuration of 5 was examined by TACu method⁹⁾. The positive contribution (Δ [M]_{436(TACu)} + 605°) of **5** suggested that this sugar belonged to the L-series. Therefore, avidinosamine in the sugar moiety of 3 was determined to be 3-amino-2,3,6-trideoxy-3-C-methyl- α -L-ribo-hexopyranose.

Hydrolysis of 3 (6 N HCl, 110°C, 18 hours) gave an aromatized aglycone, which was identified

Table 3. Physico-chemical properties of 5 (methyl α -L-avidinosaminide)^a.



	¹³ C (ppm)	¹ H (ppm) in CD ₃ OD
 1	98.8	4.71 (br d, $J=3.5$) ^b
2	, 39.0	2.00 (d, $J = 15$),
		1.86 (dd, J = 15, 3.5)
3	56.6	
4	75.0	3.24 (d, J=10)
5	65.5	3.66 (dq, J = 10, 6.5)
6	18.3	1.18 (d, $J = 6.5$)
3-CH ₃	24.8	1.23 (s)
1-OCH ₃	55.6	3.23 (s)

- ^a MW 175 (FAB mass, m/z 176 (M+H)⁺), $[\alpha]_{D^4}^{26}$ (c 0.085, MeOH) +103°, $[\alpha]_{435}^{24}$ (c 0.04, H₂O) +62.1°, $[\alpha]_{435}^{24}$ (c 0.04, tetraamminecopper (II) sulfate) +408°, Δ [M]_{TACu} +605°.
- ^b Multiplicity and coupling constant (Hz) are in parentheses.

Table 4. ¹³C NMR chemical shifts of methyl 3-C-methyl-hexopyranosides (δ ppm in CDCl₃).



	R ₁ : R ₂ :	NHAc ^a CH ₃	CH ₃ ^a NHAc	NO2 ^b CH3	CH3 ^b NO2	NHOH° CH ₃	NHOBz ^c CH ₃	NHBz ^c CH ₃
1		99.7	99.7	98.6	99.1	97.8	97.5	98.1
2		38.5	40.5	41.7	41.4	38.4	39.7	41.8
3		56.7	56.4	89.4	89.3	57.4	59.2	55.6
4		87.9	82.9	77.0	74.6	80.0	78.9	78.6
5		69.7	70.3	71.1	70.5	65.2	62.4	63.2
6		18.4	19.0	18.3	18.3	18.2	17.6	17.5
3-CH ₃		24.7	20.0	25.2	18.1	23.3	23.9	24.0
1-OCH ₃				56.5	56.5	55.0	55.3	55.4

^{a~c} Data taken from refs 7, 6, and 10, respectively.

		4	D	ecilorone ^a
1	152.3		153.0	
2	131.1		131.3	
3	125.6	7.49 (d, $J=8$) ^b	125.6	7.49 (d, $J=8$)
4	125.6	8.15 (d, J=8)	125.6	7.91 (d, $J = 8$)
4a	133.6		133.7	
5	173.3		173.7	
5a	106.5		106.4	
6	169.5		168.7	
6a	129.6		129.2	
7	118.4	7.88 (d, $J = 8$)	118.4	7.72 (d, $J = 8$)
8	133.9	7.55 (dd, $J = 8, 1$)	134.1	7.51 (dd, $J = 8, 1$)
9	143.5		143.6	
10	125.4	8.10 (d, J=1)	125.5	8.01 (d, $J = 1$)
10a	130.8		131.0	
11	169.0		168.4	
11a	106.5		106.6	
12	176.3		176.2	
12a	119.4		119.5	
13	21.8	2.53 (s)	22.1	2.58 (s)
1'	96.5	5.83 (d, $J=3$)	97.3	5.95 (d, $J=4$)
2'	67.0	4.32 (dd, J=11, 3)	68.4	4.24 (dd, J = 10.5, 4)
3'	63.1	2.65°	62.7	2.49 (dd, $J = 10.5, 3.2$)
4′	73.4	4.08 (br s)	75.4	3.96 (d, J=3.2)
5'	75.8		76.4	
6'	21.8	1.68 (s)	22.4	1.80 (s)
3'-N(CH ₃) ₂	42.9	2.65 (s)	43.3	2.58 (s)

Table 5. Comparison of NMR chemical shifts of 4 and decilorone (δ ppm, CDCl₃-CD₃OD (9:1))

^a Data taken from ref 2.

^b Multiplicity and coupling constant (Hz) are in parentheses.

^c J value could not be determined.

Fig. 7. Fragment ions observed in the FAB mass spectra of avidinorubicin and 3.



to be decilorone³) by comparison of physico-chemical properties (MW 491, FAB mass 492 (M+H)⁺, $[\alpha]_D^{24} + 850^\circ$ (c 0.1, CHCl₃-MeOH, 9:1), mp 190~193°C (dec)) and NMR spectral data with an authentic sample (Table 5). Since the chemical shifts and coupling constants of the dimethyl aminosugar moiety (1'-H to 6'-H and C-3'-N(CH₃)₂) directly fused to the aglycone are similar to those of arugomycin but different from those of nogalamycin¹⁰, the relative configuration of this aminosugar moiety was also confirmed to be 3,6-dideoxy-3-dimethylamino-galactopyranose with axial orientations for 2'-H ($J_{1',2'}=3$ Hz and $J_{2',3'}=11$ Hz) and 3'-H ($J_{2',3'}=11$ Hz and $J_{3',4'}=1.5$ Hz) and an equatorial orientation for 4'-H ($J_{3',4'}=1.5$ Hz).

The glycosidation shift observed at C-4' (δ 73.4 in 4 vs. 80.9 in 3) indicated the sugar chain was attached to the C-4' position. In the ¹H and ¹³C NMR spectra of 3, a pair of signals assignable to the avidinosamine unit were observed (1"-H to 6"-H and 3"-CH₃, and 1""-H to 6"'-H and 3"'-CH₃ in Table 2) and NOEs were detected between 4'-H and 1"-H, and 4"-H and 1"'-H in the NOESY spectrum of 3. These results indicated the presence of glycosidic bonds between 1" and 4', and 1"' and 4". The configurations of these glycosidic bonds were determined to be β based on their large coupling constants $(J_{1",2"}=J_{1'',2''}=9.0 \text{ Hz})$.

As a result, the structure of avidinorubicin was determined to be 1, a compound with two amino groups in place of two nitro groups in decilorubicin. The fragmentation patterns observed in positive FAB mass spectrum of 1 and 3 (Fig. 7) also supported this structure. The determination of the absolute stereochemistry of the aglycone is being investigated.

Biological Properties

1 inhibited thrombin-, ADP- or collagen-induced aggregation in human platelet with $IC_{50}s$ 7.9, 7.9, and 3.0 μ M, respectively. This compound showed cytotoxicity against HeLa cells with IC_{50} value being 1.37 μ g/ml, which was slightly weaker than those of arugomycin and decilorubicin. This may mean that the nitro or amino group in decilonitrose and avidinosamine do not play an important role in their biological activities. Detailed biological data will be reported elsewhere.

Discussion

Avidinorubicin, containing two units of a unique aminosugar was determined to be a novel anthracycline related to decilorubicin, and its aminosugar was determined to belong to the L-series.

On the assumption that the nitro group was formed biosynthetically from the amino group, avidinorubicin may be a precursor of decilorubicin. The decilonitrose moiety of decilorubicin had been determined to belong to the L-series by TACu method and synthetic approach by IsHII *et al.*⁷⁾. On the other hand, the decilonitrose moiety of viriplanins was recently claimed to be of the D-series by analysis of the CD spectra of a dibenzoate derivative of the reduced product of decilonitrose by KIND *et al.*¹¹⁾. More recently, the L-configuration of decilonitrose was established unambiguously by synthesis of its methyl β -glycoside starting from L-rhamnose by NISHIMURA *et al.*¹²⁾. It is, therefore, gratifying that the aminosugar moiety obtained from avidinorubicin also belongs to the L-series, supporting the L-configuration of the decilonitrose moiety in decilorubicin and arugomycin.

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References

- WILEY, P. F.; R. B. KELLY, E. L. CARON, V. H. WILEY, J. H. JOHNSON, F. A. MACKELLAR & S. A. MIZSAK: Structure of nogalamycin. J. Am. Chem. Soc. 99: 542~549, 1977
- SHIRLING, E. B. & D. GOTTLIEB: Cooperative description of type strains of *Streptomyces* V. Additional descriptions. Int. J. Syst. Bacteriol. 22: 265~394, 1972
- ISHII, K.; Y. NISHIMURA, H. NAGANAWA, S. KONDO & H. UMEZAWA: The structure of decilorubicin. J. Antibiotics 37: 344~353, 1984
- KAWAI, H.; Y. HAYAKAWA, M. NAKAGAWA, K. FURIHATA, H. SETO & N. ÖTAKE: Arugomycin, a new anthracycline antibiotic. II. Structural elucidation. J. Antibiotics 40: 1273 ~ 1282, 1987
- 5) HÜTTER, K.; E. BAADER, K. FROBEL, A. ZEECK, K. BAUER, W. GAU, J. KURZ, T. SCHRÖDER, C. WÜNSCHE, W. KARL & D. WENDISCH: Viriplanin A, a new anthracycline antibiotic of the nogalamycin group. I. Isolation, characterization, degradation reactions and biological properties. J. Antibiotics 39: 1193~1204, 1986
- ISHII, K.; S. KONDO, Y. NISHIMURA, M. HAMADA, T. TAKEUCHI & H. UMEZAWA: Decilorubicin, a new anthracycline antibiotic. J. Antibiotics 36: 451 ~ 453, 1983
- ISHII, K.; Y. NISHIMURA, S. KONDO & H. UMEZAWA: Decilonitrose and 4-O-succinyl-L-diginose, sugar components of decilorubicin. J. Antibiotics 36: 454~456, 1983
- SATO, K.; M. MATSUZAWA, K. AJISAKA & J. YOSHIMURA: Branched-chain sugars. XIX. On the application of ¹³C NMR spectroscopy to the configurational assignment of 3-C-substituents of aldohexopyranose derivatives. Bull. Chem. Soc. Jpn. 53: 189~191, 1980
- UMEZAWA, S.; T. TSUCHIYA & K. TATSUTA: Studies of aminosugars. XI. Configurational studies of aminosugar glycosides and aminocyclitols by a copper complex method. Bull. Chem. Soc. Jpn. 39: 1235~1243, 1966
- 10) KAWAI, H.; Y. HAYAKAWA, M. NAKAGAWA, K. FURIHATA, K. FURIHATA, A. SHIMAZU, H. SETO & N. ŌTAKE: Arugomycin, a new anthracycline antibiotic. I. Taxonomy, fermentation, isolation and physico-chemical properties. J. Antibiotics 40: 1266~1272, 1987
- 11) KIND, R.; K. HUTTER, A. ZEECK, K. SCHMIDT-BÄSE & E. EGERT: Viriplanin A, a new anthracycline antibiotic of the nogalamycin group. II. The structure of a novel hydroxyamino sugar from reduced viriplanin A. J. Antibiotics 42: 7~13, 1989
- NISHIMURA, Y.; K. ISHII & S. KONDO: The absolute configuration of decilonitrose, a sugar component of decilorubicin, is undoubtedly 2,3,6-trideoxy-3-C-methyl-3-nitro-L-ribo-hexopyranose. J. Antibiotics 43: 54~61, 1990